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Terry C. Jones

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 which is more important molality or molarity both molality  
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 interchangeably

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 solution of hcl is 1 0585 g ml the molality of the solution is 1  
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 which can be abbreviated as molar or text m m pronounced

molar the molar concentration of the solute is sometimes abbreviated by putting square brackets around the chemical formula of

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temperature which of these changes a molality b fraction of solute present in water

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