

Mother And Son Shower

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119 0023 g mol 156 g KBr

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 C_6H_{14} d 0 660 g cm^3 when it is
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 cm^3 in a 35 0 55 0 ml ratio assume that
the volumes are additive

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web notice that the mole fraction has
no units on it and is written as a
decimal value do not change it to
percent note of caution you could see
the term mole percent it is simply the
mole fraction multiplied by 100 for
example in the problem just below the
mole fraction of cinnamic acid is 0 2885
its mole percent would be 28 85

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