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N Colangelo

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2 $3\text{CaOH} + 2\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{AlOH}$ 3 type of reaction double displacement
3 $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement
4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction combustion
5 $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$ type of reaction

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