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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h<sub>2</sub> o l the spectator ions in koh aq and hno<sub>3</sub> aq are k aq and no<sub>3</sub> aq how to identify and list the

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a net charge of zero is electrically neutral

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web the ag aq and cl aq ions become agcl s but the na aq ions and the no<sub>3</sub> aq ions stay as na aq ions and no<sub>3</sub> aq ions these two ions are examples of spectator ions ions that do nothing in

the overall course of a chemical reaction

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excluding spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction the questions are intended to test the students understanding of the meaning of the reactions

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be  $2\text{Cl}^- + \text{Cu}^{2+} \rightarrow \text{CuCl}_2$  these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

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