

# Jirai La Voir Un Jour

R Barnett

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accepts an h ion a proton in an acid  
base reaction  $\text{nh}_3 + \text{h}_2\text{o} \rightleftharpoons \text{nh}_4^+ + \text{oh}^-$  here  
 $\text{nh}_3$  brønsted lowry base  $\text{h}_2\text{o}$

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the right because  $\text{hcl}$   $\text{pka} = 7$  is a  
stronger acid than  $\text{nh}_4^+$   $\text{pka} = 9.3$  and  
 $\text{nh}_3$  is a stronger base than  $\text{cl}^-$   $\text{pka} = 7.7$   
to identify the

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the following questions in the space  
provided a  $\text{h}_2\text{so}_4$  b  $\text{h}_2\text{so}_3$  c  $\text{h}_2\text{s}$  d  $\text{hclo}_4$   
e hydrogen cyanide which if any of the  
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would look solid liquid or in the form of  
gas it would also have a sour taste  
bases would feel slippery and solid in  
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web log  $\text{h} = 10.405891 \times 10^{-5} \text{ mol dm}^{-3}$   
log  $\text{pka} = 3.7540503010030050$   
and  $\text{hcooh} = 0.1$  so  $\text{hcoo} = 0.200 \text{ mol dm}^{-3}$   
amount vol x

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part c while part d about the ph of  
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