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**Richard Bailey**

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web a rate law shows how the rate of a chemical reaction depends on reactant concentration for a reaction such as  $\text{a} \rightarrow \text{b}$  products the rate law generally has the form  $\text{rate} = k [\text{a}]^n$

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web oct 27 2022 the rate law for the reaction  $\text{ce h}_2 \text{g ce } 2\text{no g ce n}_2\text{o g ce h}_2\text{o g}$  nonumber has been

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web among the lanthanides in addition to 3 states 2 states are shown by  $\text{nd sm eu tm}$  and  $\text{yb}$  only whereas 4 state is exhibited by  $\text{ce pr nd tb}$  and  $\text{dy}$  elements rest five elements show only 3 states oxidation states of lanthanide elements it has been shown that the

lanthanide elements are highly electropositive and form essentially ionic

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web to separate the lanthanides from other elements occurring with them they are chemically combined with specific substances to form lanthanide compounds with low solubility oxalates and fluorides for example a process known as ion exchange is then used to separate the lanthanides from each other in this process a solution of the **list of elements in the lanthanide series** thoughtco - Mar 23 2022

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Pm

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