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H Kauffman

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$\text{H}_2\text{O} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{H}_2\text{O}$ if it takes 27.4 ml of 0.768 M NaOH to titrate 16.7 ml of H_2SO_4 what is the concentration of the H_2SO_4 solution? Hint: balance the equation first! Q2

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