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Jin-Ying Zhang

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bertambah dari atas ke bawah sifat fisik golongan viiia gas mulia sulit bereaksi bersifat inert karena konfigurasi elektronnya stabil sehingga jarang ditemukan dalam bentuk senyawa gas mulia sedikit larut dalam air kecuali helium dan neon karena ukuran atomnya terlalu kecil

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sifat sifat gas mulia wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik molekul yg terdiri dari 1 atom ini erat kaitannya dengan struktur elektron oktet dan duplet dari gas mulia

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dec 3 2015 gas mulia dan halogen merupakan unsur kimia yang sifatnya sangat bertolak belakang gas mulia dinamakan mulia karena seperti logam mulia sukar sekali bereaksi sementara halogen halos garam artinya adalah pembentuk garam halogen memiliki 6

bilangan oksidasi sehingga mudah sekali berreaksi dengan unsur lainnya gas mulia noble

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pembuatan gas mulia gas helium mempunyai titik didih yang sangat rendah yaitu 268 8 0c sehingga pemisahan gas helium dari gas alam dilakukan dengan cara pengembunan liquefaction sampai gas alam akan mencair sekitar 156 0c dan gas helium terpisah dari gas alam pembuatan ne ar kr dan xe dilakukan dengan proses destilasi bertingkat

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golongan gas mulia dan halogen beberapa golongan unsur memiliki nama khusus seperti golongan viiia gas mulia dan viia halogen golongan tersebut memiliki nama khusus seperti sifat priodik sifat fisik dan sifat kimia kekhas sifat ke 2 golongan tersebut dapat membantu kita unsur unsur yang lain dalam 1 golongan semakin ke atas dan dalam

mengenal macam macam sifat dan

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semua unsur pada golongan ini berwujud gas dan memiliki konfigurasi elektron yang sangat stabil sehingga akan sangat sulit untuk bereaksi dengan unsur lainnya unsur unsur gas mulia antara lain adalah helium he neon ne argon ar krypton kr xenon xe dan radon rn

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nov 26 2015 1 wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik ini erat kaitannya dengan struktur elektronoktet dan duplet dari gas mulia sedangkan wujud gas pada suhu kamar disebabkan titik

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helium he neon

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