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ceramic catalyst approximated as 1 kg of ceramic per 2 kg of solution thus 3 kg kg of ceramic the reactor volume is  $12 \text{ blocks} \times 5 \text{ kg ceramic block} = 0.003 \text{ m}^3$  kg ceramic  $4 \times 0.72 \text{ m}^3$   
 answer problem 13 12 continued 1 d the reactor must contain a hot solution of sodium

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