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 web the equation $\text{mg hcl mgcl}_2 \text{ h}_2$ is
 clearly unbalanced because on one side
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$\text{HCl} + \text{Ag}_2\text{S} \rightarrow \text{H}_2\text{S} + 2\text{AgCl}$
 $\text{H}_2\text{O} + \text{HgO} \rightarrow \text{Hg} + \text{O}_2$
 $\text{Fe} + \text{CO}_3 \rightarrow \text{FeO} + \text{CO}_2$
 3 $\text{H}_2 + 3\text{O}_2 \rightarrow 3\text{H}_2\text{O}$
 write balanced chemical equations for each of the following descriptions of a chemical reaction
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 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $\text{Pb} + \text{OH}^- \rightarrow 2\text{HCl} + 2\text{H}_2\text{O} + 1\text{PbCl}_2$
 $\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2$
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web in the foreword of the third edition it is stated this edition of the iec 60601 1 2 was revised to structurally align it with the 2005 edition of iec 60601 1 and to implement the decision of iec subcommittee 62a that the clause

numbering structure of collateral

standards written to iec 60601 1 2005
would adhere to the form specified in

iso