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9.5 free energy and equilibrium

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 reaction $2 no_2 \rightleftharpoons 2 no + o_2$ for which
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observed rate $k [no_2]^2$ the following
 mechanism has been proposed
 $2 no_2 \rightleftharpoons 2 no + 2 o_2$
 $23 no_2 \rightleftharpoons 2 no + 2 o_2$

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 once the reactant has changed into the
 product it stays that way we can
 assume that eventually the reaction will
 finish when the limiting reactant runs

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