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method used to express the amount of substance this has been experimentally proving that one gram atom of any element as well as one gram molecule of any substance contains the same amount of entities the experimentally decided number is found to be 6.022137×10^{23}

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of units in a mole also bears the name avogadro s number or avogadro s constant in honor of the italian physicist amedeo avogadro
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 web the mole was made the seventh si base unit in 1971 by the 14th cgpm
 2019 redefinition of si base units before the 2019 redefinition of the si base units the mole was defined as the amount of substance of a system that contains as many elementary entities as there are atoms in 12 grams of carbon 12 the most common isotope of carbon