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web thermal energy is the thing that object possess but heat is a process that occurs between objects if we compare an iceberg to a pot of boiling water each of the molecules of both objects have their own amount of thermal energy all pieces of matter even very cold ones have at least some thermal energy because they are in motion at the

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flows into or out of the object you can use the following equation to calculate the change in thermal energy change in thermal energy j mass kg change in temperature c specific heat $j/kg \cdot c$ $q = m \cdot c \cdot \Delta T$ in the equation q stands for the change in thermal energy

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