

# Palais Des Ths V2

## Michal Rosen-Zvi

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which states that an acid is a substance  
that forms hydrogen ions h when  
dissolved in water and a base is a  
substance that forms hydroxide ions oh  
when dissolved in water

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relevant section of the specification  
need to know that bases are chemicals  
which react with acids in neutralisation  
reactions they are usually metal oxides  
metal hydroxides or metal carbonates  
examples are copper oxide and copper  
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substance that is capable of donating a  
proton hydrogen ion to another  
substance a base is a molecule or ion  
able to accept a hydrogen ion from an  
acid acidic substances are usually  
identified by their sour taste

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are not all dissociated into ions in  
aqueous solution is a weak acid or a  
weak base solutions of weak acids and  
weak bases reach a chemical  
equilibrium between the un ionized  
form of the compound and the  
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