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 the process by which nuclei emit
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 isotope that has an unstable nucleus
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 particle a positively

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 emits radiation and is transformed into

the nucleus of one or more other
 elements the resulting daughter nuclei
 have a lower mass and are lower in
 energy

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 reactions during the life cycle of a
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 lifetime of a star a different fuel is used
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 of hydrogen to give helium is the
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 and particles emitted from a
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 unstable isotopes or radioisotopes each
 alpha particle contains two protons and
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 far

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 decay spontaneously by emitting alpha
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 fission reactions caused by neutron
 releases from each reaction nuclear
 pertaining to the central point group or
 mass about which a gathering or
 concentration takes place electric
 current the net movement of electric
 charges in a single direction tracer
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 atom changes composition b
 radioactivity spontaneous emissions
 accompanying changes in nuclei of
 atoms c protons are positively charged
 and hence repel each other so the
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 chemistry is the study of radiation from
 an atomic or molecular perspective
 including elemental transformation and
 reaction effects as well as physical
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 the nuclei of unstable isotopes called
 radioisotopes gain stability by
 undergoing changes 25 1 radioactivity
 an unstable nucleus releases energy by
 emitting radiation during the process of
 radioactive decay 25 1 the three main
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