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web thermal energy from the hotter object to the cooler object the hotter object will cool down and the cooler one will warm up until they are at the same temperature where there will be no net energy transfer two objects are

in thermal equilibrium when there is no net transfer of thermal energy between them solids liquids and gases

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web thermal energy is the thing that object possess but heat is a process that occurs between objects if we compare an iceberg to a pot of boiling water each of the molecules of both objects have their own amount of thermal energy all pieces of matter even very cold ones have at least some thermal energy because they are in motion at the

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these atoms and molecules will keep
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energy contained within a system that
is responsible for its temperature heat
is the flow of thermal energy heat is the
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present in a system in a state of
thermodynamic equilibrium by virtue of

its temperature thermal energy cannot
be converted to useful work as easily as
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loosely in various contexts in physics
and engineering generally related to
the kinetic energy of vibrating and
colliding atoms in a substance it can
refer to several different well defined
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internal energy we use the first law of
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