

Jeremy Buendia Ebook

Marco Cascella

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web use the information in this chart to answer the following questions 1 if i have 125 grams of water at a temperature of 25.0 °C and heat it to 145.0 °C what will the change in enthalpy be 2 sketch the heating curve temperature vs time for the process in 1.3 i have 18.7 grams of an unknown liquid chemical at a temperature of 35.0 °C

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ml of 1.00 M HCl at 25.0 °C is added to 25.0 ml of 1.00 M NaOH at 25.0 °C in a foam cup calorimeter a reaction occurs and the temperature rises to 32.0 °C calculate the enthalpy change in kJ for this reaction assume the densities of the solutions are 1.00 g/ml and that **6.4 calorimetry chemistry libretexts** - Jan 31 2023

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