

Ejercicios De Uniones Quimicas Ionicas Y Covalentes

Gabriela Mohina, Patricia Moreno, María Gabriela Muñoz

Química Cuaderno de Ejercicios Jara Reyes, Silvia, 2014-08-01 Aprender Ciencias no sólo significa procesar un gran volumen de información actual, sino también comprender y actuar, con conocimiento de causa y con responsabilidad, en la solución de problemas que se presentan en la vida cotidiana. Para lograrlo, es necesario desarrollar una serie de habilidades intelectuales como aprender a clasificar, organizar, seleccionar, planificar el trabajo, etcétera. En este sentido, el Cuaderno de Ejercicios de Ciencias 3, Química: proporciona al estudiante un espacio para el desarrollo de habilidades y la autoreflexión acerca de su propio proceso de aprendizaje. Es un cuaderno práctico, conciso y enriquecedor. Incluye suficientes y variados ejercicios acerca de los contenidos del tercer curso de Ciencias, así como un proyecto relacionado con la temática estudiada. Contiene algunas propuestas para la realización de proyectos. Apoya el planteamiento de los contenidos con imágenes claras y precisas. Al final de cada bloque se incluye un conjunto de ejercicios que le permitirán al alumno evaluar lo aprendido. En el Cuaderno de Ejercicios de Ciencias 3, Química el alumno hallará actividades muy diversas relacionadas con toma de decisiones, juegos de palabras, interpretación de esquemas, modelos, tablas y gráficas, ejercicios de comprensión lectora, elaboración de historietas y folletos, así como ejercicios de jerarquización, clasificación, identificación y aplicación.

Química en ejercicios Programa UBA XXI // María José Billorou et al. Universidad de Buenos Aires, 2014-03-01
Ejercicios de química para estudiantes de UBA XXI

Problemas y ejercicios de aplicación para química Gabriela Mohina, Lilia Davel, 2020-09-01 Este libro brinda un compilado de problemas que responden a los temas del programa de Química que se dicta en UBA XXI. Su resolución requiere la aplicación de los conceptos fundamentales adquiridos en el estudio teórico de la materia, tales como estructura atómica y tabla periódica, uniones químicas y propiedades de las sustancias, soluciones, magnitudes atómico-moleculares y estado gaseoso, reacciones químicas, equilibrio químico, equilibrio ácido-base, soluciones reguladoras (buffer) y química orgánica. Las autoras crearon un abanico de situaciones problemáticas y de ejercitaciones con el propósito de favorecer los procesos de construcción en esta disciplina

Química en apuntes Gabriela Mohina, Patricia Moreno, María Gabriela Muñoz, 2020-03-01 Ligado directamente a la asignatura Química que se dicta en el Programa de Educación a Distancia de la Universidad de Buenos Aires (UBAXXI), este material se propone ofrecer a los estudiantes herramientas conceptuales que los auxilien en el recorrido por los conceptos

fundamentales asociados a esta materia.

Química 1. Fundamentos ,

Curso de química general Francisco Santamaría Herrero,1966

Química I Luis Enrique Pereyra,2022-10-01 Este libro te ayudará a construir los mejores aprendizajes y herramientas para que los apliques dentro y fuera del aula, proporcionándote así una mejor calidad de vida y un excelente desarrollo personal y profesional.

Química: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry.For anyone interested in learning more about chemistry.

Guía bibliográfica para el ingreso a las universidades ,1977

Física preuniversitaria. II Paul A. Tipler,1992 Este libro se ha escrito para estudiantes que cursan la asignatura de Física en la enseñanza preuniversitaria. En él se desarrolla la física elemental necesaria para futuros estudiantes de Biología, Medicina, Física, Química, Ingeniería, etc... Los te

Conceptos y modelos para comprender la química Darío Estrin,Sara Aldabe Bilmes,Luis Ignacio

Domenianni,2024-03-01 Podemos definir a la química como una ciencia natural, es decir que emplea metodologías científicas para estudiar algunos fenómenos o propiedades naturales, a los que llamamos fenómenos o propiedades químicas. Los fenómenos y propiedades químicas están determinados por las propiedades de la materia estudiada, pero además tratan de conciliar la visión atómico-molecular del fenómeno con el comportamiento macroscópico que resulta de la misma sustancia. Este libro plantea un recorrido por los conceptos básicos para descubrir conjuntamente cómo pensamos y razonamos desde los átomos, las moléculas y sus interacciones con el mundo que nos rodea. Condensa, con una fuerte carga de trabajo experimental, las bases conceptuales de la Química, con especial énfasis en la comprensión de casos sencillos para describir sistemas complejos. La vida, los minerales o las estrellas, el medio ambiente, requieren miradas disciplinares diversas, una de las cuales es la molecular y termodinámica. Con estas bases también se encaran los desafíos sociales y tecnológicos:

diseñar y fabricar nuevos materiales, nuevos fármacos, nuevos procesos sustentables, monitorear y remediar la contaminación existente, preservar la vida en nuestro planeta.

Física preuniversitaria. Volumen II Paul Allen Tipler, 2021-04-07 Este libro se ha escrito para estudiantes que cursan la asignatura de Física en la enseñanza preuniversitaria. En él se desarrolla la física elemental necesaria para futuros estudiantes de Biología, Medicina, Física, Química, Ingeniería, etc... Los temas ordinarios de Física se desarrollan en el orden tradicional que encontramos en la mayoría de cursos preuniversitarios: Mecánica (capítulos 2 a 10), Termodinámica (capítulo 11 a 14), Vibraciones y Ondas, incluido el Sonido (capítulos 15 a 17), Electricidad y Magnetismo (capítulos 18 a 23), Óptica (capítulos 24 a 27) y Física moderna (capítulos 28 a 33). Aun cuando la ordenación de los temas es la normal, hay algunos aspectos que no suelen figurar en otros libros. En el capítulo 6, que trata de trabajo y energía, se incluye un apartado referente a energía térmica y metabolismo que relaciona el tema del capítulo con la experiencia cotidiana de los lectores. El capítulo 7 (impulso, cantidad de movimiento y centro de masa) contiene un apartado relativo a la propulsión a chorro y una descripción cualitativa del movimiento de un cohete. Hay todo un capítulo (capítulo 9) que trata de la gravedad, en el que se estudia el movimiento de los satélites y el problema del escape de la Tierra. El flujo viscoso se trata en el capítulo 10 y se escriben las ecuaciones del movimiento de un fluido, la conducción de calor y la conducción eléctrica en formas análogas a fin de poner de manifiesto sus semejanzas. El capítulo relativo al segundo principio de la Termodinámica (capítulo 14) relaciona la Entropía con la pérdida de energía disponible y con el desorden y la probabilidad. Los temas de Vibraciones y Ondas (capítulos 15 a 17) se dan a continuación de la Termodinámica y con ello se termina el primer semestre. No obstante, esta materia se podría combinar fácilmente con la Óptica (capítulos 24 a 27) y desarrollarse en clase antes o después de la Electricidad y el Magnetismo, si se creyera conveniente.

Fundamentos de Química Ralph A. Burns, 2003

Química, un proyecto de la A.C.S. American Chemical Society, 2005 Este texto de Química está desarrollado y escrito por un grupo de excelentes educadores químicos escogidos por la American Chemical Society. Poniendo un constante énfasis en el aprendizaje activo, la obra propone a los estudiantes que no lean simplemente sobre el mundo de la química, sino que lo exploren de manera personal. Química es un libro esencial para aquellos profesores que quieran incorporar a su docencia el aprendizaje basado en la investigación.

Biofísica y fisiología celular Ramón Latorre, 1996

Libros en venta en Hispanoamérica y España, 1988

Química 2,

El Enlace Covalente Julio Francisco Mata Segreda, 2004

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