

# Oh Arcontes Hice Lo Correcto

## RS Peters

*What is the formula of "magnesium hydroxide"? | Socratic* Feb 27, 2017 ·  $\text{Mg}(\text{OH})_2$  Now we know that hydroxides are salts of  $\text{HO}^-$ , and some metal ion. Now if the parent metal has an electronic configuration of 2:8:2, then there are 12 electrons, ...

What are the spectator ions in the reaction between sodium Oct 16, 2016 · The nitrate and the sodium ions.  $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{AgNO}_3(\text{aq}) \rightarrow 2\text{Ag}_2\text{CO}_3(\text{s}) + 2\text{NaNO}_3(\text{aq})$  The net ionic equation is:  $2\text{Ag}^+(+) + \text{CO}_3^{2-} \rightarrow \dots$  **Could a buffered solution be made by mixing aqueous ... - Socratic** Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered ...

**0.10 moles of sodium cyanate dissolved in 250cm<sup>3</sup> distilled** Dec 28, 2015 · The cyanate anions will react with water to form isocyanic acid, HNCO, and hydroxide anions,  $\text{OH}^-$ . This tells you that you can expect the pH of the resulting solution to be ...

What is the product of the following reaction? 1)CH<sub>3</sub>OH - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write....  $\text{NH}_4^+ + \text{HO}^- \rightarrow \text{NH}_3(\text{aq}) + \text{H}_2\text{O}(\text{l})$  For methanol, the acid base reaction would proceed ...

**Question #627ef - Socratic** functional groups like -OH, -CHO, -CO- are present in carbohydrates. commonly two functional are present in a carbohydrate, -OH and -CHO, or -OH and -CO-. Aldose shows different reactions ...

What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a  $\text{M}^+$  ion. Hydroxide anion,  $\text{OH}^-$ , has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ...

What is the mass percent of hydrogen for C<sub>3</sub>H<sub>5</sub>OH? | Socratic Jan 2, 2017 · In order to compute the mass percent of hydrogen in  $\text{C}_3\text{H}_5\text{OH}$  we're going to need two things: 1: The formula mass of the entire compound 2: The atomic mass of hydrogen The ...

**If the pH=9 for a solution of KOH(aq), what is the ... - Socratic** Aug 18, 2016 · If the pH = 9 for a solution of  $\text{KOH}(\text{aq})$ , what is the concentration of  $\text{H}^+$ ? Question #44ec4 - Socratic That's a silly way to say carbon dioxide!  $\text{CO}_2(\text{g}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{CO}_3(\text{aq})$  May be wrong, but it seems reasonable. The gas would interact ...

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