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**Ensheng Dong**

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web examine the pairs of ternary acids in model 2 that contain sulfur phosphorus and nitrogen each pair has

one acid that ends in  $\text{ic}$  and another that ends in  $\text{ous}$  these endings are related to the name of the polyatomic anion found in the acid  $\text{ate}$  or  $\text{ite}$   
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hydrofluoric acid  $\text{HClO}_3$  answer chloric

acid chlorate  $\text{H}_2\text{SO}_3$  answer sulfurous

acid sulfite  $\text{H}_2\text{SO}_4$  answer sulfuric acid

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h which immediately combines with a  
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ion h<sub>3</sub>O<sup>+</sup> the h<sub>3</sub>O<sup>+</sup> ion is what defines the  
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