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 of atomic orbitals on adjacent atoms  
 and contain a maximum of two  
 electrons differences  $\sigma$  bonds are  
 stronger and result from end to end  
 overlap and all single bonds are  $\sigma$   
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 atoms are weaker because they result  
 from side by side  
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 dipole moment of zero and therefore  
 does not significantly interact with the

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 a strong electrolyte thus it should  
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