

# Padi Advanced Open Water Manual Pdf

**Scott C. Dulebohn**

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can be hydrolyzed to yield myristic acid

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composed of myristic acid and glycerol and makes up 20 25 studocu chem

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introduction in this experiment

extraction technique will be used to

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experimental light brown nutmeg

powder 4 g was mixed with 25 ml of

clear and colorless diethyl ether solvent

and the brown cloudy mixture was

filtered by gravity filtration with fluted

fast flow filter paper in a powder funnel

into a 250 ml round bottom flask rbf

leaving light brown residue and beige

filtrate diethyl ether was

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web trimyristin in the major

triglyceride found in nutmeg

representing 20 25 of the dried weight

of ground nutmeg to extract the

trimyristin we performed a solid liquid

extraction using the organic solvent

methylene chloride ch<sub>2</sub>cl<sub>2</sub>

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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyristin is a triglyceride what are triglycerides prelab question 2 how do microwaves heat microwave procedure add approximately 2.5 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh *isolation of trimyristin from nutmeg*

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point *lab 4 extraction of trimyristin from nutmeg university of toronto* - Sep 07

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web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

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