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 $\text{mg s} + 2 \text{ hcl a q} \rightarrow \text{mgcl}_2 \text{ a s} + \text{h}_2 \text{ g}$ the reaction between mg s and hcl a q is represented by the equation above in a kinetics experiment a 0.080 g sample of mg s ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M hcl a q at 25 °C
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