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compound that accepts protons from an  
acid neutral neither acidic nor basic  
pure water neutral and has a ph value  
of 7 0 ph a measure of how many  
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the greater the number of hydrogen  
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hydrogen it is a measure of how many  
excess h ions there are in a solution the  
ph scale runs from 0 to 14 with 0  
representing the highest concentration  
of hydrogen ions substances have a ph  
below 7 while alkaline substances bases  
have a ph above 7

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are on the low numbered red end of the  
ph color chart and the bases are on the  
high numbered blue green end of the  
chart the order from acidic to alkaline  
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yellow green blue or in other words d c  
a b

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ph is short for potential of hydrogen it  
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