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$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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$$\text{AgNO}_3(\text{aq}) + \text{KCl}(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{KNO}_3(\text{aq})$$

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$$\text{K}_2\text{CO}_3(\text{aq}) + 2\text{CuCl}(\text{aq}) \rightarrow$$

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1 is given as equation 3 $s_2 aq$ $fe_2 aq$ gt
 $fes s$ eq 3 8 note that the phase
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are soluble while $pbso_4$ is insoluble
sulfates are generally soluble except for
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2 cl aq 2 na aq 2 cl

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grandparents in my free time i play
football with my friends an active way
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play guitar and like to read books 3 a
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 sons and their wives and his young
 children and grandchildren on the
 death of the household head this large
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 the former high

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