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Raffaella Di Napoli

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web a nucleus changes into a new element by emitting nuclear radiations these changes are described using nuclear equations example ${}^{86}_{219}\text{Po} \rightarrow {}^{84}_{215}\text{Pb} + {}^4_2\text{He}$

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 $\text{Li} + \text{N}_2 \rightarrow \text{Li}_3\text{N}$
 $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$
 $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $\text{Cu} + \text{HNO}_3 \rightarrow \text{Cu(NO}_3)_2 + \text{H}_2\text{O} + \text{NO}$
 balance the following equations
 $\text{Fe} + \text{Cl}_2 \rightarrow \text{FeCl}_3$
 $\text{C}_4\text{H}_{10}\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$

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web sep 12 2022 the half life of the nuclei and thus the material is $t_{1/2} = 693$ million years 35 a the decay constant is $\lambda = 1.99 \times 10^{-5} \text{ s}^{-1}$

1 b since strontium 91 has an atomic mass of 90.90 g

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