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R Sandford

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$\text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$ label eq

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products are written on the right in a

balanced equation the total numbers of

atoms of each kind on both sides are

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atoms present in the reactants to form

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overall reaction is $i \rightarrow c$
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for this reaction one of the species has
been left out determine what that
species is in lieu of the question mark
make sure the charge is correct if it has
one

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considered insoluble if very little of it
dissolves in a given amount of the
solvent for example sand SiO_2 is
considered insoluble in water
nonetheless extremely small quantities
of dissolved SiO_2 can be found in a

mixture of sand and water as this
suggests solubility is a relative term
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process in which one or more
substances are changed into one or
more new substances reactants the
starting materials in a reaction
products materials present at the end
of a reaction balanced chemical
equation a chemical equation in which
the number of each type of atom is
equal on the two sides of

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involve only physical changes a sliced
orange has acid citric acid that can

react with sodium hydroxide so the
answer is d exercise 7 2 1 which of the
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directive shall apply to all federal
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any authority and execution of
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