

Apple Inc Organisational Structure

Albert A Gayle

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both compound types acids ionic

compounds with fixed charge

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can form compounds with different

ratios of their elements so prefixes are

used to specify the numbers of atoms of

each element in a molecule of the

compound examples include sf 6 sulfur

hexafluoride and n 2 o 4 dinitrogen tetroxide

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containing monatomic ions consists of

the name of the cation the name of the

metal followed by the name of the anion

the name of the nonmetallic element

with its ending replaced by the suffix

ide some examples are given in table 2

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mole of the compound would result in

two moles of bismuth a total of 417 96g

tellurium has a molar mass of one mole

of the compound would result in three

moles of tellurium a total of 382 8g add

the mass of bismuth and the mass of

tellurium per mole to find the total

molar mass divide the mass of bismuth

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compounds sodium chloride on the
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compounds state whether it is ionic or
covalent if it is ionic write the symbols
for the ions involved a nf 3 b ba₂ c nh₄
2 co 3 d sr h₂ po 4 2 e ib₂ f na₂ o
answer a answer b answer c answer d
answer e answer f problem 4 3 1 4 4 3
1 4

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only one type of atom iron atoms sulfur
is another element it contains only
sulfur atoms and nothing else so it is
pure when we mix two different pure

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2o₂ strategy first you have to know
how to name a molecular compound to
name a compound its the prefix name

of first element prefix and name of the
second element with the suffix ide the
prefixes are as follows mono 1 di 2 tri 3
tetra 4

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co₂ these compounds are very
different from ionic compounds like
sodium chloride nacl nacl

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6 d ch₃ch₂ch₂ch₂ch₃ answer a
answer b answer c answer d problem 4
4 1 3 calculate the molecular or
formula mass of each of the following a
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