

Oh Soberano Santuario Rosario

Tao Wei

What is the formula of "magnesium hydroxide"? | Socratic Feb 27, 2017 · $\text{Mg}(\text{OH})_2$ Now we know that hydroxides are salts of HO^- , and some metal ion. Now if the parent metal has an electronic configuration of 2:8:2, then there are 12 electrons, ... **What are the spectator ions in the reaction between sodium** Oct 16, 2016 · The nitrate and the sodium ions. $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Ag}_2\text{CO}_3(\text{s}) + 2\text{NaNO}_3(\text{aq})$ The net ionic equation is: $2\text{Ag}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{Ag}_2\text{CO}_3(\text{s})$... *In each of the following pairs, which one has the lower viscosity?* Jul 31, 2015 · $\text{C}_2\text{H}_5\text{Cl}$ has weak dipole-dipole forces from the C-Cl bonds, but $\text{CH}_3\text{-OH}$ has strong hydrogen bonds. Confirmation: The viscosity of $\text{C}_2\text{H}_5\text{Cl}$ is 0.279 mPa·s, while the viscosity of ... What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M^+ ion. Hydroxide anion, OH^- , has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ... *Question #93f99* - Socratic The molarity of the acid is 1.434. Start with the equation: $2\text{HNO}_3(\text{aq}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}(\text{NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ This tells us that 2 moles of HNO_3 ... *Question #44ec4* - Socratic That's a silly way to say carbon dioxide! $\text{CO}_2(\text{g}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{CO}_3(\text{aq})$ May be wrong, but it seems reasonable. The gas would interact ... *What is the product of the following reaction? 1) CH_3OH* - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... $\text{NH}_4^+ + \text{HO}^- \rightarrow \text{NH}_3(\text{aq}) + \text{H}_2\text{O}(\text{l})$ For methanol, the acid base reaction would proceed ... Could a buffered solution be made by mixing aqueous ... - Socratic Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH ? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered ... **0.10 moles of sodium cyanate dissolved in 250cm³ distilled** Dec 28, 2015 · The cyanate anions will react with water to form isocyanic acid, HNCO , and hydroxide anions, OH^- . This tells you that you can expect the pH of the resulting solution to be ... *What makes for a good leaving group?* + *Example* - Socratic Jun 23, 2015 · A good leaving group has to be able to part with its electrons easily enough, so typically, it must be a strong acid or weak base relative to other substituents on the same ...

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