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reaction to proceed transcribed image
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the questions using the graph

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reaction the energy of the products is
greater than the energy of the
reactants and Δh is positive b in an
exothermic reaction the energy of the
products is lower than the energy of the
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rate of a chemical reaction a b c d is
given in first approximation by $v = k[A]^a[B]^b$
where k depends by many factors as solvent geometry type
of reaction E_a is the activation energy R
is the universal constant of gases T is the

absolute temperature
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activation energy of a reaction and how
is this energy related to the activated
complex of the reaction E_a is the
minimum amount of energy for a
reaction to occur it is the amount of
energy required to create an activated
complex 4 what happens when a
catalyst is used in a reaction a catalyst
changes the reaction
18 4 potential energy diagrams
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that occur during a chemical reaction
can be shown in a diagram called a
potential energy diagram or sometimes
called a reaction progress curve a
potential energy diagram shows the
change in potential energy of a system
as reactants are converted into
products the figure below shows basic
potential energy
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energy that has the potential to become
another form of energy an object's

potential energy depends on its
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shown on the energy diagram above on
the right side of the energy diagram
draw a horizontal line segment to
indicate the energy of the products
draw a vertical line segment to indicate
 Δh for the reaction and label it with the
correct value see attached image
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the activation energy is not the average
kinetic energy that the colliding
particles need in order to successfully
react instead it is the energy barrier
between reactants and products often
the energy required to cross the barrier
comes from the kinetic energy of
collisions
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an

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abandons dans l histoire de l épreuve

24 heures du mans 1997 wikiwand -
Feb 08 2023

web les 24 heures du mans 1997 sont la
65e édition de l épreuve et se déroulent
les 14 et 15 juin 1997 sur le circuit de
la sarthe